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APPLICATION NO.	FILING DATE	FIRST NAMED INVENTOR	ATTORNEY DOCKET NO.	CONFIRMATION NO.
10/574,110	06/09/2006	Toshio Hayashi	2006-0436A	7170
513 7590 12/07/2007 WENDEROTH, LIND & PONACK, L.L.P.			EXAMINER	
2033 K STREET N. W.			CHO, JENNIFER Y	
SUITE 800 WASHINGTO	N, DC 20006-1021		ART UNIT	PAPER NUMBER
			1621	
			-	
			MAIL DATE	DELIVERY MODE
			12/07/2007	PAPER

Please find below and/or attached an Office communication concerning this application or proceeding.

The time period for reply, if any, is set in the attached communication.

	Application No.	Applicant(s)					
	10/574,110	HAYASHI, TOSHIO					
Office Action Summary	Examiner	Art Unit					
	Jennifer Y. Cho	1621					
The MAILING DATE of this communication appears on the cover sheet with the correspondence address Period for Reply							
A SHORTENED STATUTORY PERIOD FOR REPLY IS SET TO EXPIRE 3 MONTH(S) OR THIRTY (30) DAYS, WHICHEVER IS LONGER, FROM THE MAILING DATE OF THIS COMMUNICATION.  - Extensions of time may be available under the provisions of 37 CFR 1.136(a). In no event, however, may a reply be timely filed after SIX (6) MONTHS from the mailing date of this communication.  - If NO period for reply is specified above, the maximum statutory period will apply and will expire SIX (6) MONTHS from the mailing date of this communication.  - Failure to reply within the set or extended period for reply will, by statute, cause the application to become ABANDONED (35 U.S.C. § 133).  Any reply received by the Office later than three months after the mailing date of this communication, even if timely filed, may reduce any earned patent term adjustment. See 37 CFR 1.704(b).							
Status							
1)⊠ Responsive to communication(s) filed on <u>01 October 2007</u> .							
3) Since this application is in condition for allowan							
closed in accordance with the practice under Ex parte Quayle, 1935 C.D. 11, 453 O.G. 213.							
Disposition of Claims							
4) Claim(s) 1-9 is/are pending in the application.							
4a) Of the above claim(s) is/are withdrawn from consideration.							
5) Claim(s) is/are allowed.							
6)⊠ Claim(s) <u>1-9</u> is/are rejected.							
7) Claim(s) is/are objected to.							
8) Claim(s) are subject to restriction and/or	8) Claim(s) are subject to restriction and/or election requirement.						
Application Papers							
9)☐ The specification is objected to by the Examiner.							
10) ☐ The drawing(s) filed on is/are: a) ☐ accepted or b) ☐ objected to by the Examiner.							
Applicant may not request that any objection to the drawing(s) be held in abeyance. See 37 CFR 1.85(a).							
Replacement drawing sheet(s) including the correction is required if the drawing(s) is objected to. See 37 CFR 1.121(d).							
11) The oath or declaration is objected to by the Examiner. Note the attached Office Action or form PTO-152.							
Priority under 35 U.S.C. § 119							
<ul> <li>12) Acknowledgment is made of a claim for foreign priority under 35 U.S.C. § 119(a)-(d) or (f).</li> <li>a) All b) Some * c) None of:</li> <li>1. Certified copies of the priority documents have been received.</li> <li>2. Certified copies of the priority documents have been received in Application No.</li> <li>3. Copies of the certified copies of the priority documents have been received in this National Stage application from the International Bureau (PCT Rule 17.2(a)).</li> <li>* See the attached detailed Office action for a list of the certified copies not received.</li> </ul>							
Attachment(s)							
1) Notice of References Cited (PTO-892)  4) Interview Summary (PTO-413)							
Paper No(s)/Mail Date  Notice of Draftsperson's Patent Drawing Review (PTO-948)  Information Disclosure Statement(s) (PTO/SB/08)  Paper No(s)/Mail Date  Notice of Informal Patent Application							
Paper No(s)/Mail Date <u>7/31/07, 10/1/07</u> . 6) Other:							

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## Detailed Action

This office action is in response to Applicant's communication filed on 10/1/07. Claims 1-9 are pending in this application.

A request for continued examination under 37 CFR 1.114, including the fee set forth in 37 CFR 1.17(e), was filed in this application after final rejection. Since this application is eligible for continued examination under 37 CFR 1.114, and the fee set forth in 37 CFR 1.17(e) has been timely paid, the finality of the previous Office action has been withdrawn pursuant to 37 CFR 1.114. Applicant's submission filed on 10/1/07 has been entered.

## IDS

The information disclosure statements (IDS) filed on 7/31/07 and 10/1/07 are in compliance with the provisions of 37 CFR 1.97. Accordingly, the information disclosure statements are being considered by the examiner.

## Claim Rejections – 35 USC 103

The following is a quotation of 35 U.S.C. 103(a) which forms the basis for all obviousness rejections set forth in this Office action:

(a) A patent may not be obtained though the invention is not identically disclosed or described as set forth in section 102 of this title, if the differences between the subject matter sought to be patented and the prior art are such that the subject matter as a whole would have been obvious at the time the

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invention was made to a person having ordinary skill in the art to which said subject matter pertains. Patentability shall not be negatived by the manner in which the invention was made.

Claims 1-3, 5 and 7-8 are rejected under 35 U.S.C. 103(a) as being unpatentable over Herron et al. (WO 02/16298).

Herron et al. teaches the oxidation of xylene to both 4-methyl benzaldehyde and toluic acid, under air pressure, using a supported gold catalyst (page 13, lines 21-40, example 13 and table 1; page 7, lines 16-20). The gold catalyst is in the form of elemental gold (page 3, lines 17-18) and metal particles (page 3, line 33), on a solid support (page 3, lines 16-17), in which Group VIII metals can also be used (page 3, lines 18-20).

Herron et al. is deficient in the sense that it does not explicitly use oxygen molecules in the oxidation reaction.

However, it is well known in the art that air, which contains oxygen, can be used for oxidation reactions in lieu of pure oxygen.

Therefore, it would be prima facie obvious to one of ordinary skill in the art at the time of the invention, to use air instead of pure oxygen in the oxidation reaction for converting an aromatic compound to the corresponding aldehyde and carboxylic acid. The expected result would be the efficient formation of both an aromatic aldehyde and an aromatic carboxylic acid in good yield for the chemical industry.

Claims 1-9 are rejected under 35 U.S.C. 103(a) as being unpatentable over Herron et al. (WO 02/16298), in view of Ishii et al. (US 5,958,821).

Herron et al. teaches the oxidation of xylene to both 4-methyl benzaldehyde and toluic acid, under air pressure, using a supported gold catalyst (page 13, lines 21-40,

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example 13 and table 1; page 7, lines 16-20). The gold catalyst is in the form of elemental gold (page 3, lines 17-18) and metal particles (page 3, line 33), on a solid support (page 3, lines 16-17), in which Group VIII metals can also be used (page 3, lines 18-20).

Herron et al. is deficient in the sense that it does not teach the further conversion of the aromatic aldehyde with a primary alcohol to the aromatic carboxylic acid ester.

Ishii et al. teaches the method of oxidation of an aromatic compound having an alkyl substituent (see abstract; column 8, lines 57-58), by oxidizing it with oxygen into an aldehyde (column 9, lines 21-23) in the presence of Ag or Au (column 13, line 5), supported on a carrier (column 16, lines 18-21). The group VIII (group 8) elements (column 13, line 2) can also be further supported on the catalyst (column 16, lines 18-21). Ishii et al. further teaches the oxidation to produce an aromatic carboxylic ester (see abstract; column 9, lines 30-36), by the addition of a primary alcohol e.g. methanol, ethanol (column 29, line 2). Thus Ishii et al. teaches that oxidizing an aromatic aldehyde, using a primary alcohol as the solvent would give the corresponding aromatic carboxylic ester (column 29, lines 1-26). Furthermore, oxidation of aldehydes to carboxylic esters is old in the art.

Therefore, it would be prima facie obvious to one of ordinary skill in the art at the time of the invention, to use the teaching of Ishii et al. for the oxidation of Herron et al., to further oxidize the aromatic aldehyde to the carboxylic ester using a primary alcohol. The expected result would be the efficient formation of both an aromatic aldehyde,

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aromatic carboxylic acid and an aromatic carboxylic ester in good yield for the chemical

industry.

Any inquiry concerning this communication or earlier communications from the

examiner should be directed to Jennifer Y. Cho whose telephone number is (571) 272

6246. The examiner can normally be reached on 9 AM - 6 PM.

If attempts to reach the examiner by telephone are unsuccessful, the examiner's

supervisor, Yvonne Eyler can be reached on (571) 272 0871. The fax phone number

for the organization where this application or proceeding is assigned is 571-273-8300.

Information regarding the status of an application may be obtained from the

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system, call 800-786-9199 (IN USA OR CANADA) or 571-272-1000.

Jennifer Cho Patent Examiner

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J. PARSA PRIMARY EXAMINER

Yvonne Eyler

Supervisory Patent Examiner

**Technology Center 1600**